

Thermodynamics Multiple Choice

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Thermodynamics Multiple Choice

Thermodynamics Multiple choice Questions :-1. Which of the following variables controls the physical properties of a perfect gas (a) pressure (b) temperature (c) volume (d) all of the above (e) atomic mass. Ans: d. 2. Which of the following laws is applicable for the behavior of a perfect gas (a) Boyle's law (b) Charles'law (c) Gay-Lussac law

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Thermodynamics Multiple Choice Questions (MCQ) and Answers ...

This set of Thermodynamics Multiple Choice Questions & Answers focuses on "Maximum Work in a Reversible Process". 1. For a process from state 1 to state 2, heat transfer in a reversible process is given by a) Q for reversible= $(T_0)(S_1-S_2)$ b) Q for reversible= $(T_0)(S_2-S_1)$ c) Q for reversible= $(T_0)/(S_1-S_2)$ d) Q for reversible= $(T_0)/(S_2-S_1)$ View ...

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a $2pv-1/2\#960v^2$ b $2pv+1/2\#960v^2$ c $2pv$ d None of the above. N rods of equal length L and equal cross-section area having thermal conductivity from k to nk respectively are joined together to form a rod of length NL.

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Multiple Choice Questions on Thermodynamics - Examtime Quiz

1/16 THERMODYNAMICS Dr. Leonardo C. Medina, Jr. Multiple-Choice Test Problems Basic Concepts of Thermodynamics 1. Consider a submarine cruising 30 m below the free surface of seawater whose density is 1025 kg/m³.The increase in the pressure exerted on the submarine when it dives to a depth of 110 m below the free surface is (a) 480 kPa (b) 804 kPa (c) 1400 kPa (d) 144 kPa (e) 1100 kPa 2.

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Thermodynamics Multiple Choice Questions and Answers ...

This set of Thermodynamics Multiple Choice Questions & Answers (MCQs) focuses on "Available Energy Referred to a Cycle". 1. Which of the following is a type of energy? a) high grade energy b) low grade energy c) both of the mentioned d) none of the mentioned View Answer

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of Thermodynamics Multiple Choice Questions Answers This is the Multiples Choice Questions Part 1 of the Series in Thermodynamics as one of the General Engineering and Applied Sciences (GEAS) topic. In Preparation for the ECE Board Exam make sure to expose yourself and

Engineering Thermodynamics Multiple Choice Questions And ...

MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question. 1) A chemical reaction that absorbs heat from the surroundings is said to be ____ and has a ____ ΔH at constant pressure. A)endothermic, positive B)endothermic, negative C)exothermic, negative D)exothermic, positive E)exothermic, neutral

AP Chemistry Practice Test, Ch. 6: Thermochemistry ...

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Question 6 A drug used to treat hypertension undergoes a decomposition reaction to give an insoluble product. Calculate the temperature at which this reaction becomes spontaneous if the enthalpy of the reaction at 298 K is 51 kJ mol⁻¹ and the entropy of the reaction at this temperature is 118.74 J K⁻¹ mol⁻¹.

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