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Practice Problems

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fundamental concepts of chemistry including structure and states of matter, intermolecular forces, and reactions.

... Stoichiometry; ...  
7%-9% of exam score .

Unit 5: Kinetics You'll explore various methods to observe the changes that occur during a chemical reaction and the effects of a series of reactions.

**AP Chemistry - AP**

*Page 7/17*

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N. Acids & bases (5) O.

Gas laws & KMT (6) P.

Electrochemistry &

redox (3) P. Solutions &

concentration (6) Q.

Nuclear chemistry (5)

R. Thermo, kinetics,

equilibria (12) S.

Organic chemistry (3)

T. Solids, identifying

unknowns (3) U.

Reference tables (3)

Practice worksheets. A.

Scientific method &

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conversions (4) C ...

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**The Cavalcade o'**

**Chemistry |**

**Celebrating 20 years  
of ...**

The mole, symbol mol, is the SI base unit of amount of substance.

The quantity amount of substance is a measure of how many elementary entities of a given substance are in an object or sample. Depending on what the substance is, an

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elementary entity may be an atom, a molecule, an ion, an ion pair, or a subatomic particle such as an electron. For example, if beaker A contains 10 moles of water ...

### **Mole (unit) - Wikipedia**

The atom is the fundamental unit of all matter that makes up all living and non-living things. Build your AP® Chemistry toolkit by

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learning how to use moles to compare different units, interpret the periodic table, and examine mass and photoelectron spectra. Compare and contrast mixtures and pure substances.

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General Chemistry 1 covers all of the topics typically covered in first semester General

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Chemistry and includes both formative assessments, with high scaffolding, and end of unit and module quizzes. This course offers highly contextualized, engaging content, designed in a logical flow that transitions smoothly between relatively small amounts of expository text, worked examples, activities,

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**General Chemistry 1**  
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Instructions. To perform a stoichiometric calculation, enter an equation of a chemical reaction and press the Start button. The reactants and products, along with their coefficients will appear above.

**Reaction**  
**Stoichiometry**  
**Calculator -**

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Chemistry

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The practice of chemistry is not limited to chemistry books or laboratories: It happens whenever someone is involved in changes in matter or in conditions that may lead to such changes.

1.1.2. The Scientific Method Chemistry is a science based on observation and experimentation.

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-CHEMISTRY-module-  
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The Chemistry Primer helps students remediate their chemistry math skills and prepare for their first college chemistry course. Pre-built Assignments get students up to speed at the beginning of the course. Math is covered in the context



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of chemistry, basic  
chemical literacy,  
balancing chemical  
equations, mole  
theory, and  
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